

Synthesis of a 1,2-Diphosphaferrocene under Participation of Carbonyl Ligands via Silyl Group-shift

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The reaction of the silylated phosphinomethylenephosphane **1** with $\text{Cp}(\text{CO})_2\text{FeBr}$ gives the novel 1,2-diphosphaferrocene-system **5** (Cp = cyclopentadienyl).

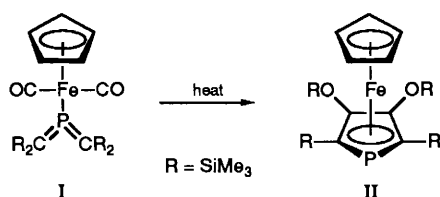
Cyclopentadienides in which threefold coordinated carbon centres are replaced by one or more phosphorus atoms play an important role in transition metal chemistry and a great number of ligands containing one¹ or more² phosphorus atoms are known.

Recently, we reported³ a novel synthetic route to phosphaferrrocenes **II** via isomerization of a ferrio-bis(methylene)phosphorane **3**, which has been obtained by reacting the silylated chloro(bismethylene)phosphorane with $[(\text{Cp})\text{Fe}(\text{CO})_2]\text{Li}$ **I** (Scheme 1). In order to get further insight into this type of CO insertion reaction with concomitant coupling of a π -bonded phosphorus(III) containing fragment, we investigated the reaction of the persilylated phosphinomethylenephosphane **1** with $(\text{Cp})\text{Fe}(\text{CO})_2\text{Br}$. With this we wanted to explore whether the reaction might be extended to other systems containing reactive elemental-silicon bonds, especially, in order to synthesise the unknown 1,2-diphospholyl ligand, in the coordination sphere of a transition metal.⁴

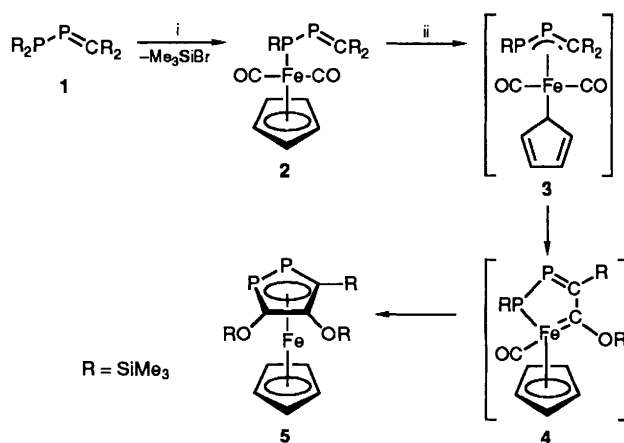
Treatment of 2.52 g (6.8 mmol) of $(\text{Me}_3\text{Si})_2\text{P}=\text{P}=\text{C}(\text{SiMe}_3)_2$ **1**, dissolved in 10 ml of toluene, with an equimolar quantity of $\text{CpFe}(\text{CO})_2\text{Br}$ and catalytic amounts of hexamethylphosphorous triamide (HMPT) immediately produced a deep red

solution. After stirring for 4 h at ca. 60 °C the product was isolated by solvent separation and fractional vacuum distillation in the form of a thermally stable, oily liquid, b.p. 137–139 °C (10^{-2} Torr) in 33% yield.

Identification of the product as the 1,2-diphosphaferrocene **5** was established by analytical and spectroscopic methods. The ³¹P NMR spectrum exhibits an AB system at high field (δ -66.2, -67.2), which is indicative of phosphorus containing cyclopentadienides.^{3,5} The value for J_{PP} (379 Hz) corresponds to a one bond coupling as is found for the sandwich complexes containing the triphospholyl ligand.



Scheme 1



Scheme 2 Reagents and conditions: i, $(\text{Cp})(\text{CO})_2\text{FeBr}$ (1 equiv.), -78 °C to room temp., toluene, HMPT; ii, 60 °C, 4 h

The presence of an unsymmetrically substituted 1,2-diphospholyl ligand is in accord with two characteristic downfield multiplets for the siloxy carbon atoms (δ 134.5, $^1J_{CP}$ 55.2, $^2J_{CP}$ 38.5 Hz; δ 131.4, $^2J_{CP}$, $^3J_{CP}$) as well as a doublet of doublets at higher field (δ 78.0, $^1J_{CP}$ 55.8, $^2J_{CP}$ 38.0 Hz)[†] in the ^{13}C NMR spectrum. The five ring carbon atoms of the cyclopentadienide ligand show a similar shielding (δ 73.5). The presence of the unsymmetrical diphospholyl ligand is also clear from the 1H NMR spectrum[‡] consisting of three chemically non-equivalent silyl groups.

The mass spectrum, with the molecular ion as the base peak, as well as the absence of a CO vibrational band in the IR spectrum are further evidence for the sandwich structure of **5**.

For the formation of **5** the following mechanism was established for the interaction of **1** with $CpFe(CO)_2Br$ (Scheme 2). ^{31}P NMR spectroscopic monitoring at $-78^\circ C$ indicated a formation of a phosphinomethylenephosphane complex **2** (δ -95.0 , 494.2 , $^1J_{PP}$ 256 Hz) in the primary

reaction step. Rearrangement should give a highly reactive η^3 -1,2-diphosphaallylic species **3**, which initiates the silyl group shift to give a carbenic type intermediate **4**. Subsequent silyl group shift from phosphorus to carbonyl oxygen produces then the final product **5**. Further investigations in this field are in progress.

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[†] Additional $^{13}C\{^1H\}$ NMR-data (20 MHz, ext. Me_4Si): **5** ($20^\circ C$, C_6D_6) δ 0.17 (dd, $^3J_{CP}$ 3.5, $^4J_{CP}$ 2.5 Hz C_3SiC), 0.76 (d, $^4J_{CP}$ 2.0 Hz C_3SiO), 1.77 (s, C_3SiO).

[‡] 1H NMR (80 MHz, int. Me_4Si): **5** ($20^\circ C$, $CDCl_3$) δ 0.03 (s, $SiMe_3$), 0.18, 0.28 (s, $OSiMe_3$), 4.42 (s, C_3H_5).